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Another type of organic molecule contains nitrogen without being, strictly speaking, an amine: carboxylic acid derivatives containing a trivalent (three-bond) ammonia in ground state are actually amides instead of amines. Amides and amines have different structures and properties, so the distinction is actually

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very important.

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23.1. Properties of amines | Organic Chemistry II

Amines are organic compounds which contain and are often actually based on one or more atoms of nitrogen. Structurally amines resemble ammonia in that the nitrogen can bond up to three hydrogens, but amines also have additional properties

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based on their carbon connectivity.

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The addition of nitrogen into an organic framework leads to two families of molecules. Compounds containing a nitrogen atom bonded in a hydrocarbon framework are classified as amines.

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Compounds that have a nitrogen atom bonded to one side of a carbonyl group are classified as amides. Amines are a basic functional group.

20.4 Amines and Amides - Chemistry

Amines Chapter - 13

Organic Chemistry 1.

AMINES NOTE:

Nomenclature and Conversions to be taught in class. 1.

Amines are regarded

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as derivatives of ammonia in which one, two or all three hydrogen atoms are replaced by alkyl group

2. Classification of amines: 3.

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- Protonation of amines with acids leads to formation of aminium salts - Aminium salts are formed from 1 o, 2 o or

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3 o amines and the
aminium ion bears at
least one hydrogen •
Quaternary ammonium
salts have four groups
on the nitrogen - The
nitrogen atom is
positively charged but
does not bear a
hydrogen atom

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constitute an important

class of organic

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compounds derived by replacing one or more hydrogen atoms of NH_3 molecule by alkyl/aryl group(s). In the IUPAC system, the amines are regarded as alkanamines, e.g.,

Structure The nitrogen atom in amine is sp^3 -hybridised.

Chemistry Notes for class 12 Chapter 13 Amines

In organic chemistry, amines (/ ə'mi:n,

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'æmi:n /, UK also /

'eɪmi:n /) are

compounds and functional groups that contain a basic nitrogen atom with a lone pair.

Amine - Wikipedia

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An organic compound A on reduction gives compound B which on reaction with chloroform and potassium hydroxide forms C. The compound C on catalytic reduction gives N-methylaniline.

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Chemistry Chapter 13, Amines belongs to the unit, Organic Compounds Containing Nitrogen. This Unit holds a total weightage of 4 marks in the final examination. Get 100 percent accurate CBSE NCERT Solutions for Class 12 Chemistry Chapter 13 (Amines) solved by expert Chemistry teachers.

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Organic

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Organic Chemistry Chapter 24: Amines, Amino Acids and ...

Amines. An amine is an organic compound that can be considered to

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be a derivative of ammonia $\left(\text{NH}_3\right)$.

Amines are molecules that contain carbon-nitrogen bonds.

Amines are named by naming the alkyl groups attached to the nitrogen atom, followed by the suffix -amine as illustrated here for a few simple examples: The name of larger molecules involve the class-identifying

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**9.9: Nitrogen-
Containing
Compounds- Amines
and Amides ...**

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The Journal of ...

Amines are derivatives of ammonia in which one or more of the hydrogens has been replaced by an alkyl or aryl group. The nomenclature of amines is complicated by the fact that several different nomenclature systems exist, and there is no clear preference for one over the others.

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